

# Sulfuric Acid, 0.5N

## Safety Data Sheet



### Section 1

### Product Description

Product Name:	Sulfuric Acid, 0.5N
Manufacturer number	SA1590-A
Distributor:	Best Sanitizers, Inc. PO Box 1360 PennValley, CA 95946
Chemical Information Emergency: Aquaphoenix Scientific	1.800.255.3924

### Section 2

### Hazard Identification

Classification of the substance or mixture:

Corrosive to metals.

Category 1

### Warning



**Appearance**—Aqueous solution

**Physical state**—Liquid

**Odor**— Odorless

### Hazard Statements

May be corrosive to metals.

### Precautionary Statements

If medical advice is needed, have product container or label at hand.

Keep out of reach of children.

Read label before use.

Keep only in original container.

Absorb spillage to prevent material damage.

# Safety Data Sheet

## Section 3

## Composition/Information on Ingredients

ChemicalName	CASNo.	Weight-%
Sulfuric Acid	7664-93-9	2.576
Purified water	7732-18-5	97.424

## Section 4

## First Aid Measures

### First Aid Measures

#### **Eye Contact**

Hold eye(s) open and rinse slowly and gently with water for 30 minutes. Remove contact lenses, if present, after first 5 minutes, then continue rinsing eye(s). Get medical advice/attention.

#### **Skin Contact**

Wash affected area with water. Rinse/flush exposed skin gently using water for 15-20 minutes. Seek immediate medical attention. Neutralize the soaking solution with sodium hydroxide solution.

#### **Inhalation**

Remove to fresh air. Seek immediate medical attention if discomfort or irritation persists.

#### **Ingestion**

Rinse mouth thoroughly. Do NOT induce vomiting.

Drink sips of water.

Seek medical attention if irritation, discomfort or vomiting persists.

### Most important symptoms and effects, both acute and delayed

Irritation. Nausea. Headache. Shortness of breath. Burning of eyes or skin. Coughing. Strong inorganic acid mists containing sulfuric acid can cause cancer. Lung damage, chronic bronchitis. Damage to teeth and stomach.

### Indication of any immediate medical attention and special treatment needed

If seeking medical attention, provide SDS document to physician. Use of soap may assist with neutralization on exposed skin in conjunction with flushing.

# Safety Data Sheet

## Section 5

## Fire-Fighting Measures

### Suitable Extinguishing Media

If in laboratory setting, follow laboratory fire suppression procedures. Use appropriate suppression agents for adjacent combustible materials or sources ignition. Use dry chemical, foam or carbon dioxide to extinguish fire.

### Unsuitable Extinguishing Media

Do not use water directly on sulfuric acid.

### Specific hazards arising from the chemical

Combustion products may include carbon oxides or other toxic vapors. Poisonous sulfur oxides are combustion products. Aerosols or mist may be produced in a fire. Sulfuric acid may ignite combustibles.

### Protective Equipment and Precautions for Firefighters

As in any fire, wear self-contained breathing apparatus pressure-demand, MSHA/NIOSH (approved or equivalent) and full protective gear.

## Section 6

## Accidental Release Measures

### Personal Precautions, Protective Equipment and Emergency Procedures

#### Personal Precautions

Use personal protection recommended in Section 8. Ensure adequate ventilation, especially in confined areas. Keep unprotected persons away. Keep away from ignition sources. Protect from heat. Neutralize with lime or soda ash. Stop the spill if possible.

### Environment Precautions

#### Environmental Precautions

Prevent entry into waterways, sewers, basements or confined areas. See Section 12 for additional ecological information.

### Methods and material for containment and cleaning up

#### Methods for containment

If in a laboratory settings, follow Chemical Hygiene Plan procedures. Always obey local regulations. Place into properly labeled containers for recovery or disposal. If necessary, use trained response staff/contractor.

#### Methods for cleaning up

Do not use water. Neutralize with lime or soda ash. Add water to slurry. Decant water to drain with excess water.

## Section 7

## Handling and Storage

### Precautions for Safe Handling

#### Advice on Safe Handling

Prevent formation of aerosols. Do not mix with bases. Wash hands after handling. Avoid contact with skin and eyes. Follow good hygiene procedures when handling chemical materials. Do not eat, drink or smoke or use personal products when using this product. Do not handle with incompatibles. Avoid inhalation of vapor or mist.

# Safety Data Sheet

## Conditions for safe storage, including any incompatibilities

Storage Conditions/	Keep Containers tightly closed in a dry, cool and well-ventilated place. Avoid storage near extreme heat, ignition sources or open flame. Keep away from food stuffs. Store with like
Incompatible materials	hazards. Protect from freezing.

## Section 8

## Protection Information

### Control Parameters:

7664-93-9 Sulfuric Acid, OSHA PEL: 1 mg/m<sup>3</sup>.  
7664-93-9 Sulfuric Acid, ACGIH TLV: 0.2 mg/m<sup>3</sup>.

### Appropriate Engineering Controls

#### Engineering Controls

Use in chemical hood only. Emergency eyewash fountains and safety showers should be available in the immediate vicinity of use/handling. Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapor or dusts below the applicable workplace exposure limits. Occupational exposure limits indicated above. Use under a fume hood. It is recommended that all dust control equipment such as local exhaust ventilation and material transport systems involved in handling of this product contain explosion relief vents or an explosion suppression system or an oxygen deficient environment. Ensure that dust-handling systems (such as exhaust ducts, dust collectors, vessels and processing equipment) are designed in a manner to prevent the escape of dust into the work area, no leakage from equipment.

# Safety Data Sheet

## Individual Protection Measures, such as personal protective equipment

Eye/Face protection	Wear safety glasses with side shields(or goggles).
Skin and body protection	Wear impermeable and resistant to the product/ substance/preparation protective gloves. Selection of glove material on considration of the penetration times, rates of diffusion and degradation.
Respiratory protection	Not required under normal conditions of use. If exposure limits are exceeded or irritation is experienced, NIOSH/MSHA approved respiratory protection should be worn. Positive-pressure supplied air respirators may be required for high airborne contaminant concentrations. Respiratory protection must be provided in accordance with current local regulations.
General Hygiene	The usual precautionary measures are to be adhered to when handling chemicals. Keep away from food, beverages and feed sources. Immediately remove al soiled and contaminated clothing. Wash face, hands and any exposed skin thoroughly after handling. Wash contaminated clothing and shoes before reuse. Do not Eat, Drink or Smoke when using this product. Do not inhale gases/fumes/dust/mist/vapors/aerosols. Avoid contact with the eyes and skin.

## Section 9

## Physical and Chemical Properties

### Information on basic physical and chemical properties

Formula:	See Section 3	Physical State:	Liquid
Odor:	Odorless	Appearance:	Aqueous solution
Odor Threshold:	Not Determined	Color:	Clear

<u>Property</u>	<u>Values</u>
pH	< 1
Meting Point/Freezing Point	Below 0°C
Boiling Point/Boiling Range	Approximately 100°C
Flash Point	Not Determined.
Evaporation rate	Not Determined.
Flammability (solid, gas)	Not Determined.
Flammability Limit in Air	
Upper flammability limit:	Not Determined.
Lower flammability limit:	Not Determined.
Vapor pressure:	Not Determined.
Vapor density:	Not Determined.
Specific Gravity	Approximately 1
Water solubility	Soluble in water.
Partition coefficient	No information available.
Autoignition temperature	Not Determined.
Decomposition temperature	Not Determined.
Kinematic viscosity	No information available.
Dynamic viscosity	No information available.

# Safety Data Sheet

## Section 10

## Stability and Reactivity Data

<b>Reactivity</b>	Reacts violently with water with evolution of heat. Corrosive to metals.
<b>Chemical Stability</b>	No decomposition if used and stored according to specifications.
<b>Possibility of Hazardous Reactions</b>	Reacts violently or explosively with incompatibles. Reacts with most metals to produce hydrogen gas, which may form explosive mixtures with air.
<b>Conditions to avoid</b>	Store away from incompatible substances, excess heat.
<b>Incompatible materials</b>	Organics. Metals. Strong acids. Strong bases. Alcohol. Chlorine. Halogenated compounds. Combustible materials. Chlorates. Alkalines. Carbides. Fulminates. Reducing agents. Nitrates. Acetic acid. Oxidizing agents.
<b>Hazardous Decomposition Products</b>	Oxides of Sulfur. Carinogenic mists/aerosols. Oxygen.

## Section 11

## Toxicity Data

### Acute Toxicity:

Oral:	7664-93-9	LD50 Rat: 2140 mg/kg
Inhalation:	7664-93-9	LD50 Rat: 510 mg/m <sup>3</sup> -2h

Sensitization	No Information Available
Germcell mutagenicity	No Information Available
Carcinogenicity	Strong inorganic acid mists containing sulfuric acid: IARC Group 1
Reproductive toxicity	No Information Available
STOT single exposure	No Information Available
STOT repeated exposure	No Information Available

# Safety Data Sheet

## Section 12

## Ecological Data

### Ecotoxicity

7664-93-9 EC50 *Daphnia magna* (Water flea) - 29 mg/l - 24 h

7664-93-8 LC50 *Gambusia affinis* (Mosquito fish) - 42 mg/l - 96 h.

### Persistence and degradability

Not applicable for test method.

### Bioaccumulation

Not Bioaccumulative.

### Mobility in soil

Aqueous solution has high mobility in soil.

### Other adverse effects

Concentrated sulfuric acid has moderate acute and chronic toxicity to aquatic life, which is driven by the pH of the aquatic environment, as a result of the presence of the acid. Small quantities will be neutralized by natural alkalinity.

## Section 13

## Disposal Information

### Waste treatment methods

Disposal of wastes

Disposal should be in accordance with applicable regional, national and local laws and regulations. Do not dispose together with household garbage. Do not allow product to reach sewage system or open water.

## Section 14

## Transport Information

DOT

UN Number 2796

Class 8 Corrosive substances

UN Name Sulfuric Acid Solution

Packing Group II

## Section 15

## Regulatory Information

### International Inventories

TSCA

Complies

DSL/NDSL

Complies

### Legend:

TSCA—United States Toxic Substances Control Act Section 8(b) Inventory

DSL/NDSL—Canadian Domestic Substances List/Non-Domestic Substances List

### US Federal Regulations

#### SARA 313

Section 313 of Title III of the Superfund Amendments and Reauthorization of 1986 (SARA).

7664-93-9 Sulfuric acid.

#### SARA 311/312

Reactive, Acute, Chronic.

#### CERCLA

# Safety Data Sheet

7664-93-9 sulfuric acid 1000 lbs.

## US State Regulations

### California Proposition 65

7664-93-9 sulfuric acid chemical known to cause cancer

## Section 16

## Additional Information

<u>NFPA</u>	Health Hazards	Flammability	Instability	Physical and Chemical Properties
	1	0	0	W
<u>HMIS</u>	Health Hazards	Flammability	Physical Hazards	Personal protection
	1	0	0	X

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### Disclaimer

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**End of Safety Data Sheet**