# **SAFETY DATA SHEET**



## 1. IDENTIFICATION

**Product identifier** 

Product Name Hydrochloric Acid, 7.7N

Other means of identification Manufacturer number: HA6207-B

<u>Distributor Address</u> **Best Sanitizers, Inc.** PO Box 1360 Penn Valley, CA 95946

**Emergency telephone number** 

**Emergency Phone Numbers** Aquaphoenix Scientific: 1-800-255-3924

## 2. HAZARDS IDENTIFICATION

## Classification

This chemical is considered hazardous by the 2012 OSHA Hazard Communication Standard (29 CFR 1910.1200).

	· ,
Specific target organ toxicity following single exposure	Category 3
Acute Toxicity (oral,dermal,inhalation)	Category 3
Serious eye damage	Category 1
Skin corrosion	Category 1A
Corrosive to metals	Category 1

#### Signal word Danger

#### **Hazard statements**

May be corrosive to metals. Causes severe skin burns and eye damage.

Causes serious eye damage.

Toxic if swallowed.

May cause drowsiness or dizziness.



Appearance Aqueous solution Physical State Liquid Odor Pungent Odor

## **Precautionary Statements**

#### **Precautionary Statements - Prevention**

Keep out of reach of children. Wear protective gloves/protective clothing/eye protection/face protection. Do not eat, drink or smoke when using this product. Avoid breathing dust/fume/gas/mist/vapors/spray. Use only outdoors or in a well-ventilated area. Read label before use. Wash skin thoroughly after handling.

#### **Precautionary Statements - Response**

If medical advice is needed, have product container or label at hand.

#### Precautionary Statements - Storage

Keep only in original container.

#### <u>Precautionary Statements - Disposal</u>

Dispose of contents/container to an approved waste disposal plant.

#### Hazards not otherwise classified (HNOC)

None.

### Other Information

No ingredients of Unknown Acute Toxicity.

## 3. COMPOSITION/INFORMATION ON INGREDIENTS

Chemical Name	CAS No.	Weight %	Trade Secret
Hydrochloric Acid, ACS	7647-01-0	75.05	
Water	7732-18-5	24.95	

<sup>\*</sup> The exact percentage (concentration) of composition has been withheld as a trade secret.

## 4. FIRST AID MEASURES

### First aid measures

**Eye Contact** Hold eye(s) open and rinse slowly and gently with water for 15-20 minutes. Remove

contact lenses, if present, after first 5 minutes, then continue rinsing eye(s). Get medical

advice/attention.

Skin Contact Wash affected area with soap and water. Rinse/flush exposed skin gently using water for 15-

20 minutes. Get medical attention if irritation persists or if concerned.

**Inhalation** Remove to fresh air. Seek immediate medical attention if discomfort or irritation persists.

Ingestion Rinse mouth thoroughly. Do NOT induce vomiting. Drink sips of water. Seek medical

attention if irritation, discomfort or vomiting persists.

#### Most important symptoms and effects, both acute and delayed

### **Most Important Symptoms and Effects**

Inhalation may cause irritation to nose and upper respiratory tract, ulceration, coughing, chest tightness and shortness of breath. Higher concentrations cause tachypnoea, pulmonary oedema and suffocation. Ingestion may cause corrosion of lips, mouth, oesophagus and stomach, dysphagia and vomiting. Pain, eye, ulceration, conjunctival irritation, cataracts and glaucoma may occur following eye exposure. Erythema and skin irritation, as well as chemical burns to skin and mucous membranes may arise following skin eposure. Potential sequelae following ingestion of hydrochloric acid include perforation, scarring of the esophagus or stomach obstruction. In some cases, RADS may develop. Respiratory symptoms may take up to 36 hours to develop.

#### Indication of any immediate medical attention and special treatment needed

If seeking medical attention, provide SDS document to physician.

#### 5. FIRE-FIGHTING MEASURES

## Suitable Extinguishing Media

If in laboratory setting, follow laboratory fire suppression procedures. Use appropriate suppression agents for adjacent combustible materials or sources of ignition.

#### **Unsuitable Extinguishing Media**

None.

#### Specific Hazards Arising from the Chemical

Combustion products may include carbon oxides or other toxic vapors. Hydrogen chloride gas.

#### Protective equipment and precautions for firefighters

As in any fire, wear self-contained breathing apparatus pressure-demand, MSHA/NIOSH (approved or equivalent) and full protective gear. Thermal decomposition can produce poisoning chlorine. Hydrochloric acid reacts also with many organic materials with liberation of heat.

## 6. ACCIDENTAL RELEASE MEASURES

### Personal precautions, protective equipment and emergency procedures

Personal Precautions Use personal protection recommended in Section 8. Ensure adequate ventilation,

especially in confined areas. Keep unprotected persons away. Keep away from ignition sources. Protect from heat. Contained spilled material by diking or using inert absorbent.

For emergency responders

**Environmental precautions** 

Environmental Precautions Prevent entry into waterways, sewers, basements or confined areas. See Section

12 for additional ecological information.

## Methods and material for containment and cleaning up

Methods for Containment Prevent further leakage or spillage if safe to do so. Contain and collect spillage with non-

combustible absorbent material, (e.g. sand, earth, diatomaceous earth, vermiculite) and place in container for disposal according to local/national regulations (See Section 13).

Methods for Cleaning Up

Use clean non-sparking tools to collect absorbed material. May be ignited by friction, heat,

sparks or flames. Collect spillage. Soak up with inert absorbent material. Sweep up and shovel into suitable containers for disposal. Following product recovery, flush area with water. Place in properly labeled containers. If in a laboratory setting, follow Chemical

Hygiene Plan procedures.

## 7. HANDLING AND STORAGE

#### **Precautions for safe handling**

#### Handling

Prevent formation of aerosols. If opening metal containers, use non-sparking tools because of the possibility of hydrogen gas being present. Wash hands after handling. Avoid contact

with skin, eves

and clothing. Do not eat, drink or smoke when using this product. Use personal protection recommended in Section 8. When handling hydrochloric acid avoid contact with metals and organic matters. Never use hot water and never add water to the acid!

#### Conditions for safe storage, including any incompatibilities

#### Storage Conditions/ Incompatible materials

Keep Containers tightly closed in a dry, cool and well-ventilated place. Avoid storage

near extreme heat, ignition sources or open flame. Keep away from foodstuffs. Store from oxidizing agents. Protect from freezing and physical damage. Containers for hydrochloric acid must be made from corrosion resistant materials: glass, polyethylene, polypropylene, polyvinyl chloride, carbon steel lined with rubber or ebonite.

## 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

#### Control parameters

#### **Exposure Guidelines**

Chemical Name	ACGIH TLV	OSHA PEL	NIOSH IDLH
Hydrochloric Acid 7647-01-0	2 ppm ceiling	TWA: 7 mg/m3	5 ppm Ceiling 7 mg/m3 Ceiling

ACGIH TLV: American Conference of Governmental Industrial Hygienists - Threshold Limit Value. OSHA PEL: Occupational Safety and Health Administration - Permissible Exposure Limits. NIOSH IDLH: Immediately Dangerous to Life or Health.

## Appropriate engineering controls

## **Engineering Controls**

Emergency eyewash fountains and safety showers should be available in the immediate vicinity of use/handling. Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapor or mists below the applicable workplace exposure limits.

#### Individual protection measures, such as personal protective equipment

Wear safety glasses with side shields (or goggles). **Eye/Face Protection** 

Wear impermeable and resistant to the product/ substance/preparation protective gloves. Skin and Body Protection

Selection of glove material on consideration of the penetration times, rates of diffusion and

degradation.

**Respiratory Protection** If exposure limits are exceeded or irritation is experienced, NIOSH/MSHA approved

> respiratory protection should be worn. Positive-pressure supplied air respirators may be required for high airborne contaminant concentrations. Respiratory protection must be

provided in accordance with current local regulations.

**Hygiene Measures** The usual precautionary measures are to be adhered to when handling chemicals. Keep away

from food, beverages and feed sources. Immediately remove all soiled and contaminated clothing. Wash face, hands and any exposed skin thoroughly after handling. Wash

contaminated clothing and shoes before reuse. Do not Eat, Drink or Smoke when using this product. Do not inhale gases/fumes/dust/mist/vapors/aerosols. Avoid contact with the eyes

and skin.

## 9. PHYSICAL AND CHEMICAL PROPERTIES

## **Physical and Chemical Properties**

See Section 3 **Formula** 

Liquid

Physical State Appearance Aqueous solution Odor Pungent odor Color Clear **Odor Threshold** 0.3 - 14.9 mg/m3

**Values Property** Remarks/ Method

рΗ < 1

Melting/freezing point Approximately 0°C None known Boiling point / boiling range Approximately 100°C None known

Flash Point Not Flammable N/A

**Evaporation rate** > 1.00 None known Flammability (solid, gas) Non-Combustible None known

Flammability Limits in Air

Upper flammability limit
Lower flammability limit
Nonexplosive
None known
Vapor pressure
No data available
Vapor density
None known
No data available
None known
None known
None known

Vapor density No data available None known **Specific Gravity** Not determined None known Water Solubility Soluble in water None known Solubility in other solvents No data available None known Partition coefficient: No data available None known **Autoignition temperature** No data available None known **Decomposition temperature** No data available None known Kinematic viscosity No data available None known **Dynamic viscosity** No data available None known

## 10. STABILITY AND REACTIVITY

#### Reactivity

Reacts violently with bases and is corrosive.

## **Chemical stability**

No decomposition if used and stored according to specifications.

## Possibility of Hazardous Reactions

Reacts violently with oxidants forming toxic gas (chlorine). Attacks many metals in the presence of water forming flammable/explosive gas (hydrogen).

#### **Conditions to avoid**

Excess heat. Incompatible products.

#### **Incompatible materials**

Metal oxides, formaldehyde. Strong bases. Most metals. Strong oxidizing agents. Reducing agents. Alkalis, cyanides, sulfides. Sulfites.

#### **Hazardous Decomposition Products**

Carbon oxides (CO, CO2). Fumes of hydrogen chloride and hydrogen in contact with metals. Oxides of carbon.

#### 11. TOXICOLOGICAL INFORMATION

#### Information on toxicological effects

Inhalation Hydrochloric acid

Sensitization No Information Available

Germ cell mutagenicity No Information Available

Carcinogenicity IARC: Group 3

Reproductive toxicity No Information Available

STOT single exposure No Information Available

STOT repeated exposure No Information Available

## 12. Ecological Data

## **Ecotoxicity**

None.

## Persistence and Degradability

Readily Biodegradable.

#### **Bioaccumulation**

Not Bioaccumulative.

#### **Mobility**

Aqueous solution has high mobility in soil.

#### Other adverse effects

None.

## 13. DISPOSAL CONSIDERATIONS

## **Disposal methods**

Disposal of wastes

Cover spill with soda ash or calcium carbonate. Mix and add water to form slurry. Decant to drain. Treat the solid residue as normal refuse. All chemical waste generators must determine whether a discarded chemical is classified as hazardous waste. Disposal should be in accordance with applicable regional, national, and local laws and regulations. Do not dispose together with household garbage.

## 14. TRANSPORT INFORMATION

DOT

UN ID Number UN1789

UN proper shipping name HYDROCHLORIC ACID

Hazard Class 8
Packing Group | |

## 15. REGULATORY INFORMATION

### **Chemical Inventories**

TSCA Complies
DSL/NDSL Complies

TSCA - United States Toxic Substances Control Act Section 8(b) Inventory DSL/NDSL - Canadian Domestic Substances List/Non-Domestic Substances List

#### U.S. Federal Regulations

#### **SARA 313**

Section 313 of Title III of the Superfund Amendments and Reauthorization of 1986 (SARA).

7647-01-0 Hydrochloric Acid

## SARA 311/312 Hazard Categories

Acute Health Hazard Yes
Chronic Health Hazard No
Fire Hazard No
Sudden Release of Pressure Hazard No
Reactive Hazard No

## **US State Regulations**

## **California Proposition 65**

None of the ingredients is listed.

## **16. OTHER INFORMATION**

NFPA Health Hazard 3 Flammability 0 Instability 1 Physical & Chemical Hazards 0

HMIS Health Hazard 3 Flammability 0 Physical Hazard 1 Personal Protection X

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## **General Disclaimer**

The information provided in this Safety Data Sheet is correct to the best of our knowledge, information, and belief at the date of its publication. The information given is designed only as a guidance for safe handling, use, processing, storage, transportation, disposal, and release and is not to be considered a warranty or quality specification. The information relates only to the specific material designated and may not be valid for such material used in combination with any other materials or in any process, unless specified in the text.

**End of Safety Data Sheet**